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The fourth edition of this key textbook provides a detailed and thorough exposition of the principles of thermodynamics, thermochemistry, thermo-kinetics, rate processes and stoichiometry, and the fundamentals of thermodynamics. It is a comprehensive and interesting treatment of theory simply as it applies to the practice of industrial, laboratory, and pharmaceutical research. How to read this book: The book is organized into four parts, and each part is divided into eight chapters. The first part deals with thermal processes in chemical systems, in particular, the heat of formation of hydrates, melting, and vaporization. In the second part, equilibrium and steady-state thermodynamic data about gases, liquids, and solids are presented. Part 3 describes the fundamentals of kinetic processes, with the focus on equilibrium and steady-state kinetics. Part 4 also describes steady-state kinetics, with applications of the topics in different fields of science and industry. Part 5, the last part, concludes the book by dealing with the issues of drug discovery, therapeutic product development, bioanalytical methods, and other topics. The text has been written to particularize the applications of thermodynamics in different fields of science and technology. It is a comprehensive and interesting treatment of thermodynamics as it applies to the practice of industry, laboratory, and pharmaceutical research. The fourth edition has been tested with thousands of students and will be supported by many lecturers. A two-year education project is being funded to support the text's dissemination and reach out to clinicians who can use it. This book is a unique collaborative work of a team of biologists, biochemists, chemists, food scientists, mechanical engineers, biophysicists, pharmacologists, physicists, and computer scientists from the University of Kinshasa (DRC), the University of Essen (Germany), the University of Karachi (Pakistan), and other research institutions. The team included a well-qualified group of scientists from academia and industry in the DRC. The team was guided and supervised by prof. Yves LeClair (University of Kinshasa, DRC) for scientific, methodologic, and editorial issues.

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Aromatic hydrocarbons are volatil... An aromatic hydrocarbon, methanol, is used as a solvent to dissolve methylene blue. A small amount of methylene blue is added to the solution of methanol containing triethylamine. The solution is stirred and the absorption spectrum is observed every 2 min. The absorption spectrum of the solution is shown in Table 4.1.2 of Pharmaceuticals: The Design and Manufacture of Medicines (4th ed, 2007, p. 106). Methanol... Aromatic hydrocarbons are volatil... [Show more] Methylene blue is an aromatic cyanin... Methylene blue is an... Aromatic hydrocarbons are volatil... [Show more] Methylene blue is an aromatic cyanin... The chapters in this edition each include a topic map to help you navigate the chapters. Interspersed throughout the text are notes (in the form of asterisks) which, when clicked, contain a list of references for further reading. If you are the owner of this book, you can access these notes and definitions straight from the book through the Web. More information is available at www.sophamedia.co.uk/ Any content on this Web Site may be screened, recorded, monitored, accessed, retained, copied, stored, disclosed, or distributed in any form or by any means whatsoever, including without limitation electronic, mechanical, photocopying, recording, scanning, transmission, facsimile, telephone, email or any other form or mechanism now known or hereafter developed, for any purpose whatsoever, including, but not limited to, use in partial or entirely different form, without the prior written permission of Aultons Pharmaceuticals. Sec8ef588b

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