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the da vinci code - information about the da vinci code, dan brown's novel, the history of da vinci code, the free download of the da vinci code ebook, the connection of the da vinci code with the halloween film, the da vinci code review and so on. Ascorbic acid and non-heme iron interact synergistically in yeast. Natural, live yeast must be aerated to grow, but vigorous aeration causes substantial accumulation of hydrogen peroxide. The addition of ascorbic acid (AsA) reduces the accumulation of hydrogen peroxide and protects the yeast from oxidative damage. Iron can act as a catalyst for the formation of hydrogen peroxide. In many species of micro-organisms, certain strains of yeast and plant cells, AsA is synthesized de novo via the ascorbate pathway and is coupled to the synthesis of ascorbate peroxidase (APX). A strain of yeast that does not express APX is shown to accumulate AsA when AsA is supplied exogenously. When yeast is incubated simultaneously with AsA and non-heme iron, the level of intracellular AsA, measured as the water-soluble oxalate complex, increases as the concentration of iron increases from 0 to 1 microM. The AsA-iron interaction is independent of the oxygen concentration, and is demonstrated even under anaerobic conditions. However, the addition of APX to the incubation medium causes a decrease in the level of AsA-iron interaction. HPLC analysis of yeast shows that under aerobic conditions, a substantial part of the AsA in yeast is oxidized to dehydroascorbate. In contrast, under anaerobic conditions, most of the AsA present in yeast appears to be present as the water-soluble oxalate complex. The implications of these data with respect to the role of AsA in protecting mammalian cells from oxidative damage are discussed.

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